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Novel Analogues of Tiazofurin by Lawesson Reagent Effected Cyclization

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**NOVEL ANALOGUES OF TIAZOFURIN
BY LAWESSON REAGENT EFFECTED CYCLIZATION**

Bożenna Golankiewicz* and Piotr Januszczyk

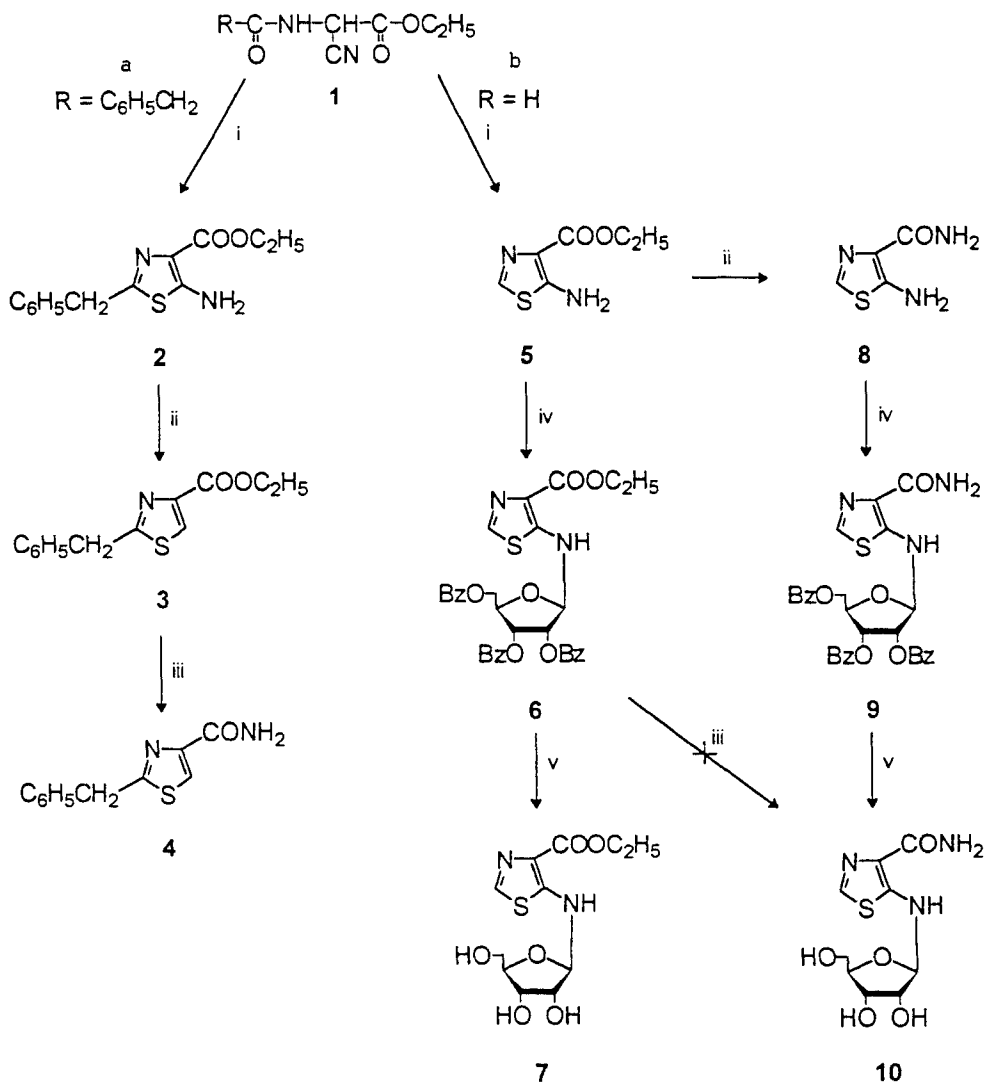
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ABSTRACT: 2-Benzylthiazole-4-carboxamide **4** and 5-(β -D-ribofuranosylamino)thiazole-4-carboxamide **10** were synthesized from phenylacetylamino- and formylamino cyanoacetic acid esters **1a** and **1b**, respectively. The ribosylation reaction leading to **10** gave rise also to its α anomer as a minor product.

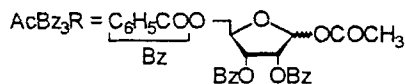
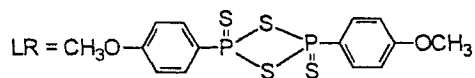
Tiazofurin (2- β -D-ribofuranosylthiazole-4-carboxamide) is a potent antitumor and antiviral agent. The compound acts *via* the inhibition of inosine monophosphate dehydrogenase (IMP DH) forming the corresponding NAD-like tiazofurin adenine dinucleotide after phosphorylation at the 5'-hydroxyl group.^{1,2} The previous alterations in its structure, which aimed at the enhancement of its selectivity and understanding of the mechanism of action, led to a conclusion that the presence of thiazole-4-carboxamide unit³ and the ribose moiety⁴ is indispensable for its biological activity.

We were interested in two types of novel tiazofurin analogues. One was 2-benzylthiazole-4-carboxamide **4**, tiazofurin with ribose moiety replaced by a group known to bind to a hydrophobic area on some enzymes.⁵ The other was 5-(β -D-ribofuranosylamino)thiazole-4-carboxamide **10**, a glycosylamine carrying in different arrangement two crucial structural units mentioned above.

The synthetic route we used is shown in the scheme.



Reagents and conditions : i, LR, benzene, reflux; ii, n-pentyl nitrite, THF, anh, reflux; iii, NH_3/MeOH 10M in ampule, 120°C , 48 h; iv, BSA, CH_3CN , reflux, 12h, then AcBz_3R , TMSTf reflux; v, $\text{Et}_3\text{N}/\text{MeOH}$ 1:10 v/v, room temperature.



The starting aminothiazoles **2** and **5** were prepared from the appropriate acylaminocynoesters **1** subjected to Lawesson Reagent (LR) in refluxing benzene according to the cyclization reaction we reported earlier.⁶ Compound **2** (mp 156-157°C, 30% yield) was then deaminated with *n*-pentyl nitrite in refluxing anhydrous tetrahydrofuran to give **3** (mp 77-78°C, 57%). The aminolysis of **3** performed with 10 M methanolic ammonia at 120°C provided the target 2-benzylthiazole-4-carboxamide **4** (mp 195-196°C, 90%) identical in all respects with the compound isolated upon photochemical degradation of cephalosporins.⁷

The synthesis of *N*-arylglucosylamines by coupling amines to reducing sugars in organic solvents has been found to afford the pyranose isomers rather than the furanose isomers.^{8,9} Therefore, we employed the condensation of the silylated aminothiazole **5** with 1-*O*-acetyl-2,3,5-tri-*O*-benzoyl- β -D-ribofuranose in refluxing acetonitrile in the presence of trimethylsilyl triflate (TMSTf, 0.27 eq). Thus, the protected 5-(ribofuranosylamino)thiazole-4-carboxylic acid ethyl ester **6** was formed (colorless oil, 52%). An attempt to remove the benzoyl groups with 10 M NH_3/MeOH at room temperature resulted in the cleavage of the glycosidic bond. The removal was accomplished by the treatment of **6** with $\text{Et}_3\text{N}/\text{MeOH}$ (1:10, v/v) at room temperature for 72 h to afford **7** (colorless solid, 79%).

Due to the instability of the *N*-exoamino glycosidic bond in concentrated methanolic ammonia it was impossible to convert compound **6** directly into the final 5-(β -D-ribofuranosylamino)thiazole-4-carboxamide **10**. This goal was reached by performing the aminolysis of the 4-substituent at the stage of aglycone. Thus, **5** treated with 10 M NH_3/MeOH at 120°C provided **8** (mp 136°C, 85%) which was reacted with peracylated sugar under the conditions identical to that applied for ribosylation of ester precursor **5** to give protected **9** (sticky crystals, 44%). The removal of the benzoyl groups with $\text{Et}_3\text{N}/\text{MeOH}$ afforded the desired thiazofurin analogue **10** (colorless solid, 70%).

From ^{13}C NMR data the ribosylated compounds **6**, **7**, **9** and **10** were identified as furanosides because of the appearance of their C-4' signals at the range 79.1 - 80.5 ppm showing the proximity of ether oxygen.¹⁰ In addition, deprotected compounds **7** and **10** had deuterium exchangeable signals of 5'-hydroxyls in their ^1H NMR spectra.

The NMR data indicated that compounds **6**, **7**, **9** and **10** were mixtures of anomers. Anomeric configurations for the major and minor components of deprotected compounds were assigned on the basis of H-1' coupling constants¹¹ in the deuterium exchanged spectra. The compounds exhibiting the small $J_{1,2}$ values (3.7-4.3 Hz) were identified as α , while those showing large $J_{1,2}$ value (9.3) as β . In the target carboxamide derivative **10** the β anomer prevailed ($\beta:\alpha = 3:2$).

The N-exoamino glycosylic bond of 5-(ribofuranosylamino)thiazoles **7** and **10** was found to be stable at pH 3, 7 and 10.

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